

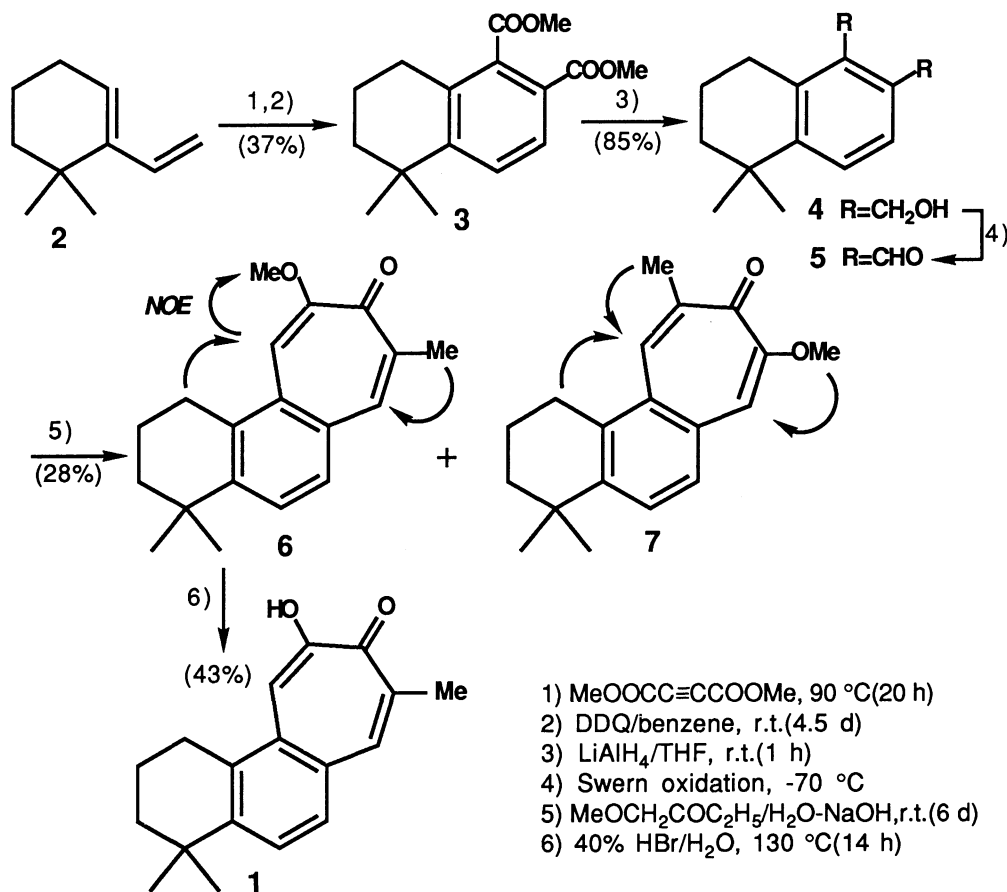
Synthesis of Salviolone, a Cytotoxic Benzotropolonoid Bisnorditerpene
from Salvia miltiorrhiza

Masahito MORI, Yoshinobu INOUE, and Hiroshi KAKISAWA*

Department of Chemistry, University of Tsukuba,
Tennodai, Tsukuba, Ibaraki 305

The structure of salviolone was unequivocally confirmed by synthesis in six steps starting from 3,3-dimethyl-2-vinylcyclohexene and methyl acetylenedicarboxylate.

Salviolone (1), isolated from a Chinese drug 'Dan-shen' (the dried root of Salvia miltiorrhiza), is the first natural norditerpene having a benzotropolonoid chromophore and has shown to exhibit a cytotoxic activity against Vero cells.¹⁾ As the structure has been characterized by only spectroscopic methods, we present here its unequivocal confirmation by chemical synthesis.



Scheme 1.

Diels-Alder reaction between 3,3-dimethyl-2-vinylcyclohexene (2)²⁾ and methyl acetylenedicarboxylate proceeded smoothly by heating the mixture to give an adduct, which without isolation was treated with DDQ in benzene at room temperature to afford a phthalic ester 3.³⁾ The ester 3 was reduced to a diol 4, which was then subjected to the Swern oxidation at -70°C to give a dialdehyde 5. As the dialdehyde was unstable, the crude 5 isolated by filtration of the reaction mixture through a short anhydrous magnesium sulfate layer was used immediately in a following step.

The suspension of the dialdehyde 5 and 1-methoxy-2-butanone⁴⁾ in an aqueous 0.05 M NaOH^{5,6)} was stirred for 6 days to afford a mixture of salviolone methyl ether 6 and its regioisomer 7 (6:7=2:1) in 28% yield from the diol 4. Both products were separated by preparative TLC (silica-gel, hexane-ethyl acetate) and were characterized by the ¹H NMR spectra and NOE experiments. The reactions in aqueous alcoholic media^{5,7)} did not proceed at all.

The major isomer 6 was heated in 40% aqueous HBr solution to give salviolone 1. The spectroscopic properties (¹H- and ¹³C-NMR, IR, and UV spectra) of the synthetic and natural salviolones were identical in all respects.

References

- 1) H.Ginda, T.Kusumi, M.O.Ishitsuka, H.Kakisawa, Z.Weijie, C.Jun, and G.Y.Zhen, *Tetrahedron Lett.*, **29**, 4603(1988).
- 2) H.Kakisawa and M.Ikeda, *Nippon Kagaku Zasshi*, **88**, 476(1967).
- 3) All new compounds except 5 gave satisfactory elemental analyses.
3: bath temp: 180-190 °C/0.3 mmHg; (CCl₄)=7.71(d,1H,J=8.6 Hz), 7.34(d,1H,J=8.6 Hz), 3.81(s,6H), 2.8-2.6(m,2H), 2.5-1.7(m,4H), 1.32(s,6H).
4: mp: 83.5-85 °C; (CDCl₃)=7.30(d,1H,J=8 Hz), 7.10(d,1H,J=8 Hz), 4.82(s,2H), 4.78(s,2H), 3.0-2.8(m,2H), 2.0-1.5(m,4H), 1.28(s,6H).
5: oil; (C₆D₆)=10.4(s,1H), 10.0(s,1H), 7.4(d,1H,J=10 Hz), 7.2(d,1H,J=10 Hz), 2.7-2.4(m,2H), 1.7-1.2(m,4H), 1.1(s,6H).
6: mp: 177-178 °C; (CDCl₃)=7.71(s,1H), 7.50(d,1H,J=8.5 Hz), 7.47(d,1H,J=8.5 Hz), 7.34(s,1H), 3.93(s,3H), 3.04(t,2H,J=6.4 Hz), 2.37(s,3H), 1.94(m,2H), 1.71(m,2H), 1.34(s,6H).
7: oil; (CDCl₃)=8.14(s,1H), 7.54(d,1H,J=8.5 Hz), 7.48(d,1H,J=8.5 Hz), 7.01(s,1H), 3.93(s,3H), 3.07(t,2H,J=6.4 Hz), 2.43(s,3H), 1.92(m,2H), 1.69(m,2H), 1.33(s,6H).
- 4) P.D.Bartlett and S.D.Ross, *J. Am. Chem. Soc.*, **70**, 926(1948).
- 5) H.Fernholz, E.Hartwig, and J.-C.Salfeld, *Justus Liebig Ann. Chem.*, **576**, 131(1952).
- 6) D.S.Tarbell and J.C.Bill, *J. Am. Chem. Soc.*, **74**, 1234(1952).
- 7) D.S.Tarbell, G.P.Scott, and A.D.Kemp, *J. Am. Chem. Soc.*, **72**, 379(1950).

(Received March 27, 1989)